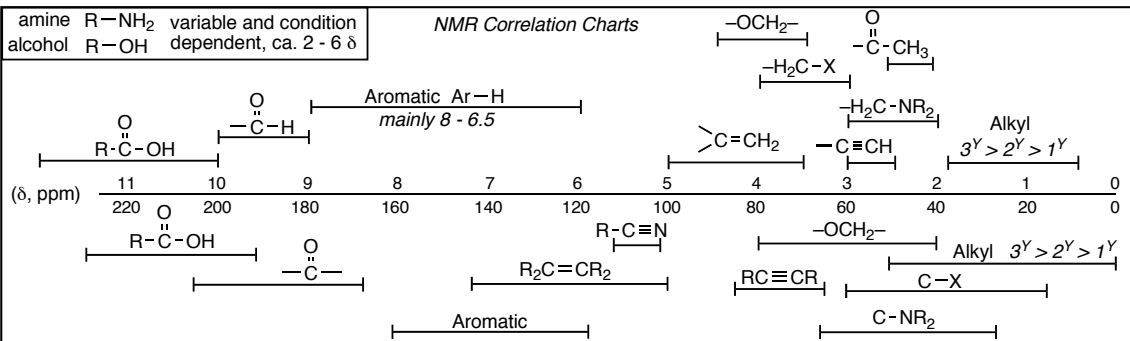
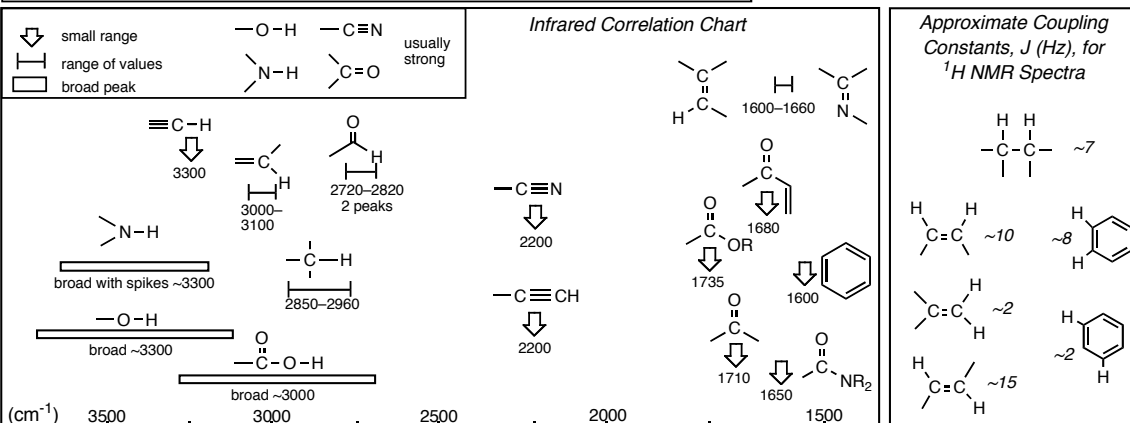


PRINTED
FIRST NAME _____PRINTED
LAST NAME _____ASU ID or
Posting ID _____Person on your **LEFT** (or Aisle)Person on your **RIGHT** (or Aisle)

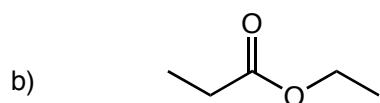
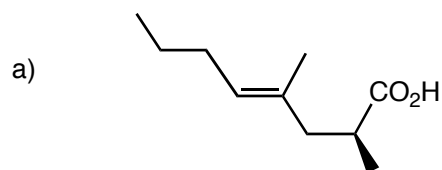
- PRINT YOUR NAME ON EACH PAGE!
- READ THE DIRECTIONS CAREFULLY!
- USE BLANK PAGES AS SCRATCH PAPER
work on blank pages will not be graded...
- WRITE CLEARLY!
- MOLECULAR MODELS ARE ALLOWED
- DO NOT USE RED INK
- DON'T CHEAT, USE COMMON SENSE!

1	/18	9	/72
2	/14	10	/22
3	/10	11	/20
4	/12	12	/40
5	/12	13	/25
6	/12	14	/50
7	/20	15	/40
8	/18		
Extra Credit		/5	Total (incl Extra)	
			/385+5	

H												He		Interaction Energies, kcal/mol																			
Li Be				B C N O F				Ne				<table style="width: 100%; border-collapse: collapse;"> <thead> <tr> <th colspan="2">Eclipsing</th> <th colspan="2">Gauche</th> </tr> </thead> <tbody> <tr> <td>H/H</td> <td>-1.0</td> <td>Me/Me</td> <td>-0.9</td> </tr> <tr> <td>H/Me</td> <td>-1.4</td> <td>Et/Me</td> <td>-0.95</td> </tr> <tr> <td>Me/Me</td> <td>-2.6</td> <td>i-Pr/Me</td> <td>-1.1</td> </tr> <tr> <td>Me/Et</td> <td>-2.9</td> <td>t-Bu/Me</td> <td>-2.7</td> </tr> </tbody> </table>		Eclipsing		Gauche		H/H	-1.0	Me/Me	-0.9	H/Me	-1.4	Et/Me	-0.95	Me/Me	-2.6	i-Pr/Me	-1.1	Me/Et	-2.9	t-Bu/Me	-2.7
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Na Mg				Al Si P S Cl				Ar																									
K Ca Sc Ti V Cr Mn Fe Co Ni Cu Zn				Ga Ge As Se Br				Kr																									
Rb Sr Y Zr Nb Mo Tc Ru Rh Pd Ag Cd				In Sn Sb Te I				Xe																									
Cs Ba Lu Hf Ta W Re Os Ir Pt Au Hg				Tl Pb Bi Po At				Rn																									



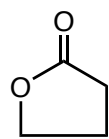
Question 1 (18 pts.) Provide IUPAC names for the following structures, do not forget to use E/Z and R/S as appropriate.



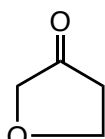
Question 2 (14 pts.) For the following three structures:

a) Clearly indicate the location of the most acidic hydrogen atom on the line-angle structures

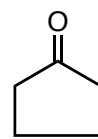
b) Indicate the order of increasing Bronsted acidity for A, B and C. Give a BRIEF explanation.



A



B



C

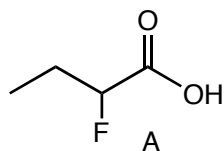
least
acidic

<

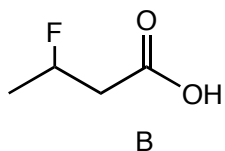
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most
acidic

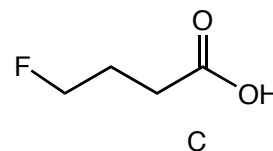
Question 3 (10 pts.) Rank the following three structures in order of increasing Bronsted acidity. Give a BRIEF explanation.



A



B



C

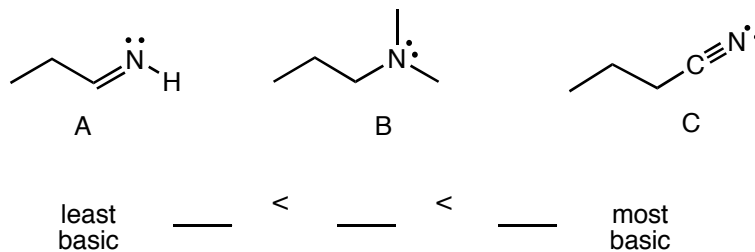
least
acidic

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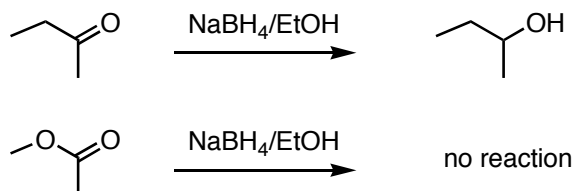
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most
acidic

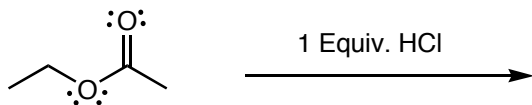
Question 4 (12 pts.) Rank the following in order of increasing basicity, give a BRIEF explanation.



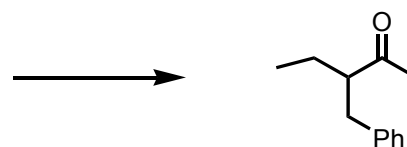
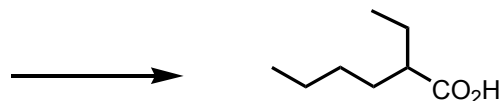
Question 5 (12 pts) Explain why sodium borohydride (NaBH_4) will reduce a ketone but will not reduce an ester. Draw minor resonance structures of the ketone and ester to support your BRIEF explanation.



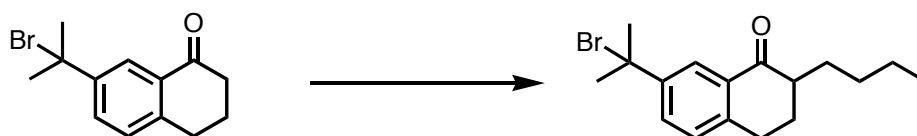
Question 6 (12 pts) Give the product of the following acid/base reaction, give a BRIEF explanation for your choice of product



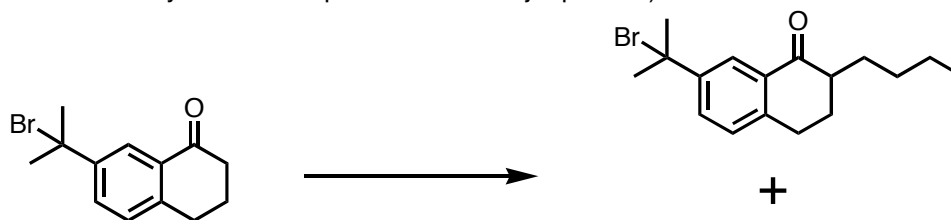
Question 7 (20 pts.) Provide the reactants that can be used to synthesize the following two structures using a malonic ester or ethylacetoacetate synthesis, i.e. give the structure of malonic ester or ethylacetoacetate and give the structures of the two bromides. You do not need to provide a reaction sequence, or any other reagents/conditions.



Question 8 (18 pts.) Give the reagents/conditions to perform the following alkylation using a Stork enamine reaction



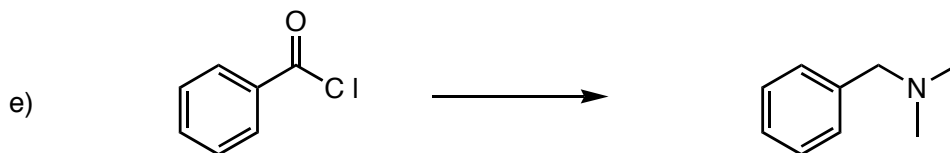
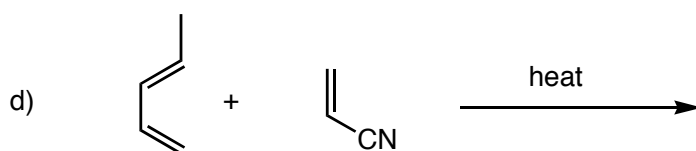
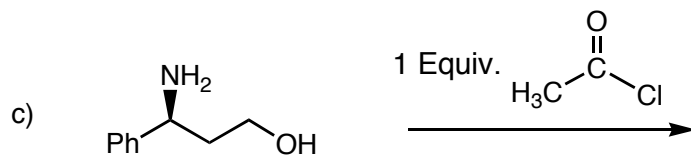
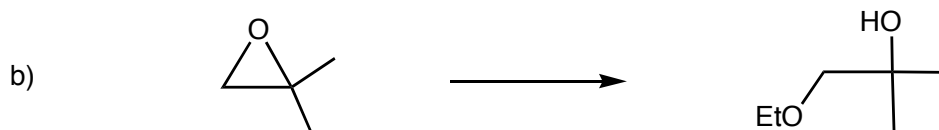
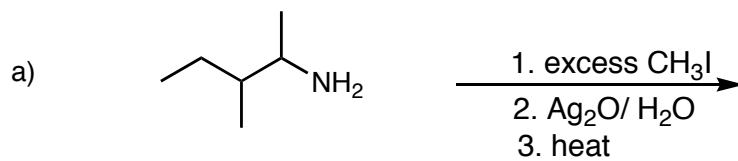
b) Give the reagents/conditions to perform the following alkylation using the LDA method, AND give the unwanted side-product that you would also expect to form under these conditions (you do not have to indicate which you would expect to be the major product).



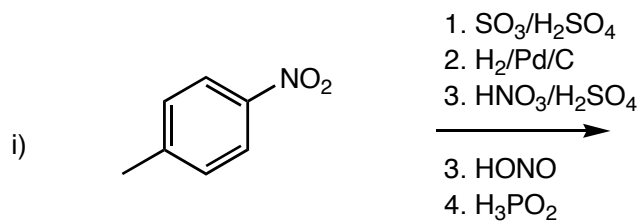
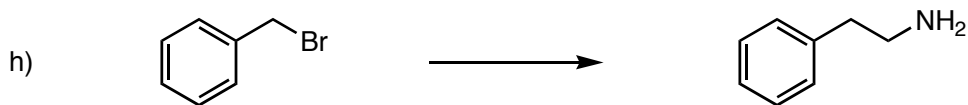
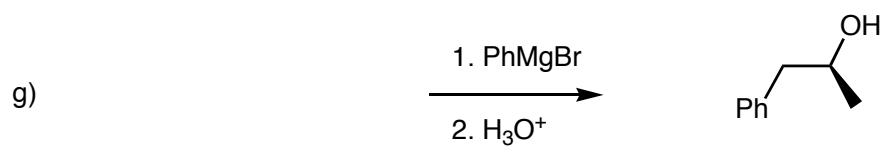
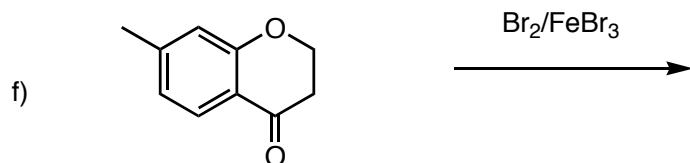
unwanted side-product

Question 9 (72 pts)

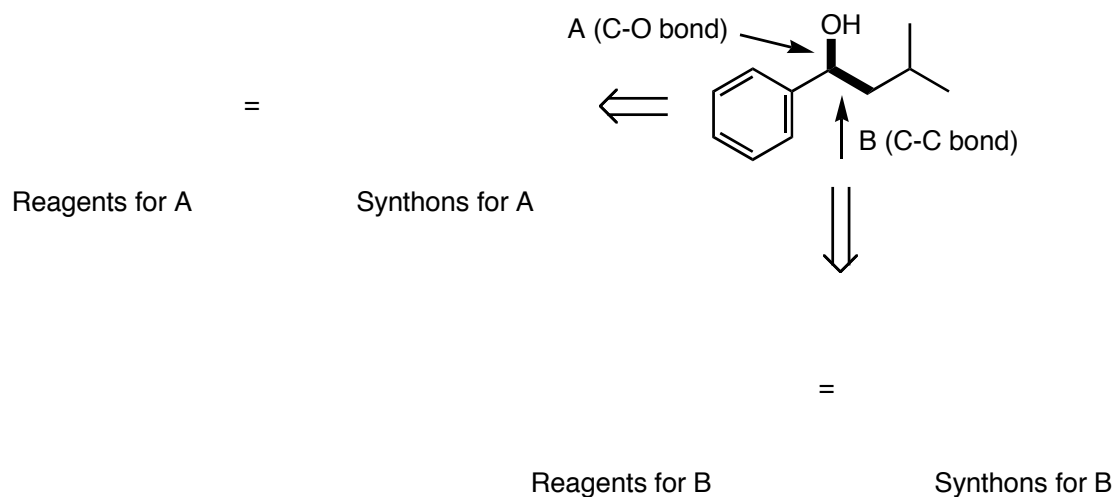
Provide the missing products, reagents/conditions or reactants, as required. **Do not forget to include stereochemistry as appropriate.**



Question 9, Contd...

Provide the missing products, reagents/conditions or reactants, as required. **Do not forget to include stereochemistry as appropriate.**

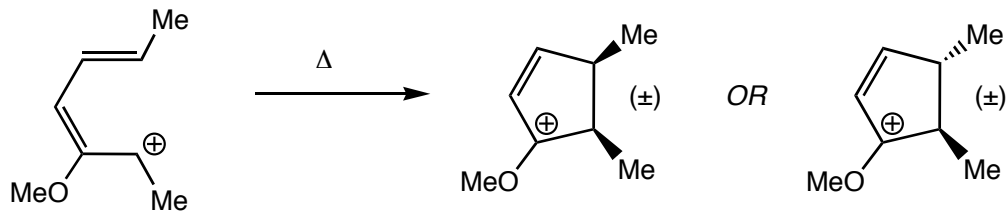
Question 10 (22 pts) For EACH of the TWO bonds indicated, A and B, provide the best SYNTHONS, and also appropriate "actual reagents"



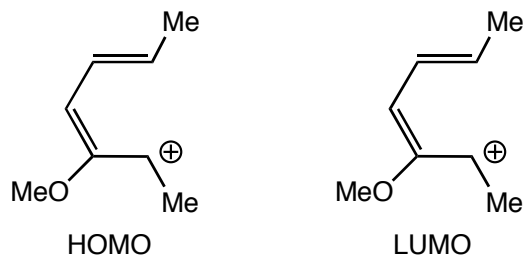
Question 11 (20 pts) For the cycloaddition reaction below:

a) Draw the curved arrow-pushing that describes product formation

b) will the stereochemistry of the expected product be cis- or trans-? Give a BRIEF explanation.

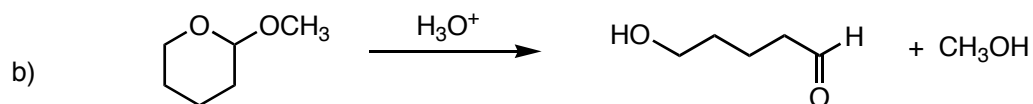
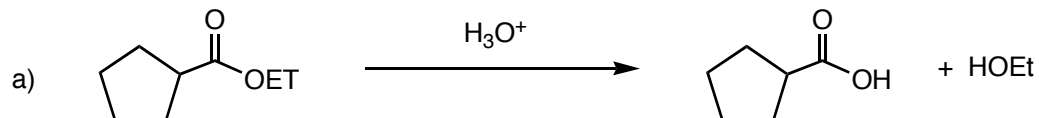


c) Draw the HOMO and LUMO of the reactant cation ON TOP of the structures that are redrawn below



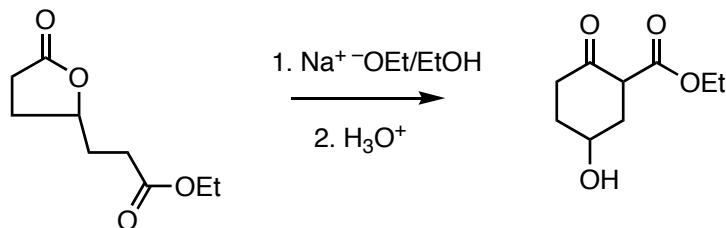
Question 12 (40 pts.) a) Give a curved arrow-pushing mechanism for the following reactions

- You can give an "abbreviated mechanism, i.e. you may use $+H^+$ and $-H^+$
- **BUT, draw all resonance structures for the intermediates**
- **Add non-bonding electrons and C-H bonds as necessary**



Question13 (25 pts.) b) Give a curved arrow-pushing mechanism for the following reaction

- **SHOW WHERE EVERY PROTON COMES FROM AND GOES TO (no $+H^+$ or $-H^+$)**
- **DO NOT DRAW RESONANCE STRUCTURES for the intermediates**
- **Add non-bonding electrons and C–H bonds as necessary**
- **At each INTERMOLECULAR step, INDICATE THE Lewis acid and base (LA or LB) and whether they are also Bronsted acids and bases (BA or BB) as appropriate**



Extra Credit Question (5 pts). Hydrolysis of which functional groups is used to make soap?

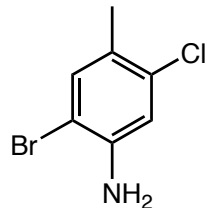
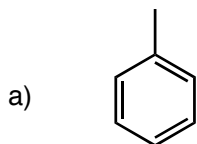
amine

ester

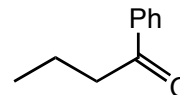
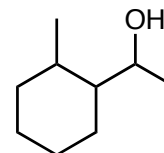
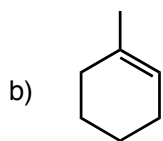
amide

aldehyde

Question 14 (50 pts.) Show how you would make the target compounds on the right from the starting compounds on the left. Show reagents and conditions where appropriate, and the structures of important intermediate compounds. Do not show any (arrow pushing) mechanisms. **For question a) you must indicate steps that require separation of isomers**

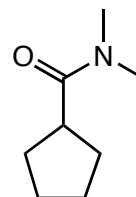


THE NEXT TWO SYNTHESIS PROBLEMS, b) and c), USE ONLY THE "SIMPLE SET OF REACTIONS" PROVIDED RECENTLY ON THE CLASS WEB PAGE!

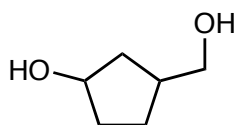


Question 15 (40 pts.) In each case, synthesize the (target) molecules on the right from the starting molecules the left. this can not be done in one reaction. Give reagents and conditions and the intermediate molecules at each step. Do not show any mechanisms or transient intermediates.

a)



b)



(ignore stereochemistry)

